



Europäisches Patentamt

European Patent Office
Office européen des brevets



1) EP 1 431 340 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication: 23.06.2004 Bulletin 2004/26

(51) Int CI.7: **C08K 5/49**, C08K 5/3492, C08L 77/00

(21) Application number: 02080410.0

(22) Date of filing: 19.12.2002

(84) Designated Contracting States:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
IE IT LI LU MC NL PT SE SI SK TR
Designated Extension States:

AL LT LV MK RO

(71) Applicant: DSM IP Assets B.V. 6411 TE Heerlen (NL)

(72) Inventor: Ottenheijm, Johannes Hendrikus Godefried 6121 NA Born (NL)

(74) Representative: Krijgsman, Willem et al DSM Intellectual Property
Office Geleen
P.O. Box 9
6160 MA Geleen (NL)

(54) Flame retardant polyamide compound

(57) The present invention relates to a flame retardant polyamide compound comprising a polyamide polymer having a weight-average molecular weight of at least 10.000 g/mol, a polyamide oligomer having a weight-average molecular weight of at most 7500 and

a halogen-free phosphorous containing flame retardant. The invention also relates to a process for preparing said flame retardant polyamide compound and to the use of said compound.

EP 1 431 340 A1

Balling S. France Troops Damin (ED

Description

5

10

15

20

25

30

45

55

[0001] The present invention relates to a flame retardant polyamide compound comprising a polyamide polymer having a weight-average molecular weight of at least 10.000 g/mol, a halogen-free flame retardant and a polyamide oligomer having a weight-average molecular weight of at most 7500, and to a process for preparing said flame retardant polyamide compound.

[0002] Such a compound is known from JP-5214246-A (Toray, 1992). JP-5214246-A describes a polyamide compound wherein the polyamide oligomer has a molecular weight of 5000 or less and is modified with hydrocarbon radicals with 5-30 carbon atoms as terminals groups, more particular the hydrocarbon radicals are attached to at least 40% or more of the terminals of the polyamide oligomer. The halogen-free flame retardant is a triazine flame retardant, in particular melamine cyanuric acid. In the process according to JP-5214246-A, the polyamide oligomer modified with said hydrocarbon radicals and the triazine flame retardant are melt-mixed with a polyamide polymer, more particular polyamide-6 and polyamide-6,6, in an amount of 0.001-10 parts by weight of the polyamide oligomer and 1-25 parts by weight of the flame retardant, relative to 100 parts by weight of the polyamide resin. The thus obtained polyamide compound is claimed to have good flame resistance, flowablility, mould separability and mechanical properties.

[0003] In many applications requiring a high flame retardancy level, an amount of 25 parts by weight of a halogen-free flame retardant, relative to polyamide polymer, is insufficient to comply with the required flame retardancy level. A disadvantage of the compounds according to JP-5214246-A is that the amount of flame retardant has to be limited to 25 parts by weight, because when said amount exceeds this limit, according to JP-5214246-A, gas is produced at the time of moulding of the compound obtained by the process, and the outer appearance of the moulded product is [0004] The object of the invention is to provide a paleage.

[0004] The object of the invention is to provide a polyamide compound comprising a halogen-free flame retardant and an polyamide oligomer, from which compound moulded products, showing a good outer appearance and having good mechanical properties can be prepared, even when the compound comprises the flame retardant in a amount larger than per the prior art.

[0005] This object is achieved by the compound wherein the halogen-free flame retardant is a halogen-free phosphorous containing flame retardant. The compound according to the invention allows the preparation of a moulded product having a good outer appearance and having mechanical properties which are at least as good as the mechanical properties of a compound prepared from a corresponding compound without polyamide oligomer. Moreover, the tardant above 25 parts by weight, more particular above 30 parts by weight, relative to the 100 parts by weight of the polyamide polymer, but even with a polyamide oligomer content above 10 parts by weight, more particular above 12 parts by weight, relative to the 100 parts by weight of the polyamide polymer.

[0006] In the context of the present invention, a polyamide compound is understood to be a polyamide composition that is obtainable by melt-mixing of the constituting components of the polyamide composition and is suitable for being used in a moulding process for preparing a shaped article. The polyamide compound may have different forms, for example, but not limited thereto, a melt (for example in an apparatus suitable for preparing moulded parts), extruded strands, chopped granules, and molded parts (for example when the polyamide composition after being melt-mixed is directly shaped into a moulded part).

[0007] In the context of the present invention a polyamide polymer is understood to be a high-molecular polyamide with a weight-average molecular weight of at least 10,000 g/mol, preferably at least 15,000 g/mol, more preferably at least 20,000 g/mol.

[0008] The polyamide oligomer is herein understood to be a low-molecular polyamide with a weight-average molecular weight of at most 7500. Preferably, the weight-average molecular weight is lower than the "molecular weight between entanglements" of the high-molecular polymer. This "molecular weight between entanglements" is for example 5,000 g/mol in the case of PA-6. Also preferably the weight average molecular weight of the polyamide oligomer is at most 5,000 g/mol, more preferably at most 4,000 g/mol, even more preferably at most 3,000 g/mol. The molecular weight of the polyamide oligomer may not be too low either, to avoid the risk of for example the glass transition temperature being lowered. Preferably the weight-average molecular weight is greater than approximately 1,000 g/mol.

[0009] Suitable polyamides, for both the polyamide polymer and the polyamide oligomer, are all the polyamides known to a person skilled in the art, comprising crystalline, semi-crystalline and amorphous polyamides that are melt-processable.

[0010] Examples of suitable polyamides according to the invention are aliphatic polyamides, for example PA-6, PA-11, PA-12, PA-4,6, PA-4,8, PA-4,10, PA-4,12, PA-6,6, PA-6,9, PA-6,10, PA-6,12, PA-10,10, PA-12,12, PA-6/6,6-copolyamide, PA-6/12-copolyamide, PA-6/11-copolyamide, PA-6,6/11-copolyamide, PA-6,6/12-copolyamide, PA-6/6,10-copolyamide, PA-6,6/6,10-copolyamide, PA-6,6/6,10-copolyamide, PA-6,6/6,10-copolyamide, PA-6,6/6,10-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,7-copolyamide, PA-6,1/6,6-copolyamide, PA-6,1/6,7-copolyamide, PA-6,1/6,7-copoly

amide, PA-6,6/6,T/6,I-copolyamide, PA-6,T/2-MPMDT- copolyamide (2-MPMDT = 2-methylpentamethylene diamine), PA-9,T, copolyamides obtained from terephthalic acid, 2,2,4- and 2,4,4-trimethylhexamethylenediamine, copolyamide obtained from isophthalic acid, laurinlactam and 3,5-dimethyl-4,4-diamino-dicyclohexylmethane, copolyamides obtained from isophthalic acid, azelaic acid and/or sebacic acid and 4,4-diaminodicyclohexylmethane, copolyamides obtained from caprolactam, isophthalic acid and/or terephthalic acid and 4,4-diaminodicyclohexylmethane, copolyamides obtained from isophthalic acid and/or terephthalic acid and/or other aromatic or aliphatic dicarboxylic acids, optionally alkyl-substituted hexamethylenediamine and alkyl-substituted 4,4-diaminodicyclohexylamine, and also copolyamides and mixtures of the aforementioned polyamides.

[0011] Preferably polyamides are chosen from the group comprising PA-6, PA-6,6, PA-6,10, PA-4,6, PA-11, PA-12, PA-12,12, PA-6,I, PA-6,T, PA-6,T/6,6-copolyamide, PA-6,T/6-copolyamide, PA-6,6,6-copolyamide, PA-6,6/6,7/6,I-copolyamide, PA-6,T/2-MPMDT- copolyamide, PA-9,T, PA-4,6/6-copolyamide, PA-6,6/6,T/6,I-copolyamides. More preferably PA-6,I, PA-6,T, PA-6,6, PA-6,6/6,T, PA-6,6/6,T/6,I-copolyamide, PA-9,T or PA-4,6, or a mixture or copolyamide thereof, is chosen as the polyamide.

10

15

20

25

30

35

40

45

50

55

[0012] The low molecular weight polyamide oligomer may be chosen to have the same composition as the high molecular weight polyamide; the two polyamides, i.e. the oligomer and the polymer, may also be chosen to have different compositions.

[0013] The high molecular weight polyamide polymer in the compound according to the invention may optionally contain modified end groups, for example amine end groups modified with a mono-carboxylic acid and/or carboxylic acid end groups modified with a mono-functional amine. Modified end-groups in the high molecular weight polyamide polymer may advantageously be applied for improved melt-stability of the compound during its preparation by melt-mixing or during a moulding process of said compound.

[0014] The polyamide oligomer preferably has a melting temperature of at least 260°C, more preferable at least 270°C, even more preferable at least 280°C. The advantage of a polyamide oligomer with a higher melting temperature, in particular when used in a larger amount, is that the high temperature mechanical properties of the polyamide compound are better retained or even improved.

[0015] A polyamide that is eminently suited for the polyamide oligomer in the process according to the invention is polyamide-4,6. Polyamide-4,6 oligomer has a melting temperature of about 290°C. The advantage of polyamide-4,6 oligomer is that it can be produced on industrial scale and that it can be combined with a wide range of high temperature engineering polyamides.

[0016] In the context of this application "PA-4,6" is understood to be a polyamide of which at least 50%, preferably at least 75%, more preferably at least 90%, consists of tetramethyleneadipamide units. PA-4,6 can be prepared through polycondensation of tetramethylenediamine and adipic acid or an adduct thereof, optionally in the presence of other polyamide-forming monomers, for example \(\epsilon\)-caprolactam, a different diamine, for example hexamethylenediamine or a different carboxylic acid, for example isophthalic acid or cyclohexanedicarboxylic acid.

[0017] The polyamide oligomer in the compound according to the invention may optionally contain modified end groups, for example amine end groups modified with a mono-functional carboxylic acid and/or carboxylic acid end groups modified with a mono-functional amine. Mono-functional carboxylic acids and/or mono-functional amines may advantageously be applied during the preparation of the polyamide oligomer as a chain stopper to control the molecular weight of the polyamide oligomer. Preferably, the polyamide oligomer comprises at most 50%, more preferably less than 40%, even more preferably at most 30% and most preferably at most 25%, modified end groups relative to the total number of end groups. The advantage of a polyamide oligomer with a lower percentage of modified end groups in the process according to the invention is that the mechanical properties of the resulting compound, when comprising a halogen-free flame retardant, are further improved.

[0018] The compound according to the invention typically comprises 0.5-30 weight % oligomer, relative to the total amount of polyamide. A person skilled in the art can choose the amount of oligomer within the given range, depending on the desired properties of the compound. Larger amounts of the oligomer can be used and may even result in a better performance of the flame retardant. However, too large amounts may adversely affect the mechanical properties. Preferably an amount of 1-20 weight % oligomer, relative to the total amount of polyamide, is chosen.

[0019] More preferably, the amount of polyamide oligomer is at most 15 weight %, even more preferably at most 11 weight %, relative to the total amount of polyamide, since a lower amount results in better mechanical properties.

[0020] Also more preferably, the amount of polyamide oligomer is at least 2 weight %, even more preferably at least 7 weight %, or even better 12 weight %, and most preferably at least 15 weight %, relative to the total amount of polyamide, since an higher amount results in a better flow.

[0021] A higher amount of polyamide oligomer is even more advantageously applied in combination with a higher amount of the halogen free phosphorous containing flame retardant. A higher content of flame retardant generally reduces the flowability of a polyamide compound. The compounds comprising these said two components in the higher amounts combine the advantageous properties of good flow and increased flame retardancy with retention of the good

mechanical properties and outer appearance.

.10

15

20

25

[0022] Suitable halogen-free phosphorus containing flame retardants that can be used in the compound according to the invention, are phosphorus-containing compounds including red phosphorus, phosphazene-based compounds, organic phosphorus compounds, and nitrogen-and-phosphorus containing compounds (such as ammonium polyphosphate and melamine based phosphorous compounds)

[0023] Suitable organic phosphorus compounds are, for example, organic phosphates, phosphines, phosphine oxides.

[0024] Examples of such phosphorus compounds are described in for example the "Encyclopedia of Chemical Technology", Vol. 10, p. 396 ff. (1980).

[0025] Many are commercially available, for example resorcinol-bis(diphenylphosphate) oligomer, under the trade name Fyrolflex® RDP from AKZO-Nobel, NL; cresyldiphenylphosphate (CDP) under the trade name Kronitex® CDP from FMC, UK; trimethylolpropanol ester of methyl phosphoric acid, under the trade name Amgard P45 from Albright and Wilson, USA; trimethylolpropanol ester of methylphosphonic acid, under the trade name Antiblaze® 1045 from Albright & Wilson, USA; polypentaerythritol phosphonate under the trade name Cyagard® RF 1041 from American Cyanamid, USA; Hostaflam® OP 910, a mixture of cyclic di- and triphosphonates from Clariant, Germany.

[0026] Suitable melamine based phosphorus compounds that can be used as the flame retardant in the inventive polyamide compounds are, for example, melamine phosphates, like melamine phosphate, melamine pyrophosphate and melamine polyphosphate, for example Melapur® 200 (DSM, the Netherlands) and PMP-100® (Nissan Chemical Industries, Japan), and phosphates of melamine condensation products, like melam polyphosphate, for example PMP-200® (Nissan Chemical Industries), and melem polyphosphate, for example PMP-300® (Nissan Chemical Industries).

[0027] In a preferred embodiment of the invention the flame retardant is a melamine based phosphorous compound. An advantage is that the inventive polyamide compound may comprise a higher melting polyamide while still retaining a good outer appearance and good mechanical properties together with retention of the flame redartant properties.

[0028] More preferable, the melamine based phosphorous is a melamine based polyphosphate. The advantage is an even better thermal stability allowing processing at higher temperature and use of a polyamide with a higher melting temperature.

[0029] Also preferably, as the flame retardant a phosphorus compound with a low volatility is chosen. It is also advantageous to use an organic phosphorus compound with a phosphorus content of at least 14 weight %, preferably

[0030] Examples of such organic phosphorus compounds with a phosphorus content of at least 14 weight % are Amgard-P45 and the metal phosphinates as for instance in US-A-4,208,321 and US-A-3,594,347. The advantage of the use of an organic phosphorus compound with a higher phosphorus content is that compounds with better mechanical properties can be obtained, while maintaining good flame retardancy performance.

[0031] In the flame-retardant polyamide compounds according to the invention the flame retardant is typically present in an amount between 1 and 100 parts by weight, relative to a total amount of polyamide of 100 parts by weight. Higher amounts may be used as well. Preferably, the amount is at least 15 parts by weight, more preferably at least 20 parts by weight, even more preferably at least 27 parts by weight and most preferably at least 40 parts by weight, relative to the total amount of polyamide of 100 parts by weight. A higher minimum amount of flame retardant is advantageously applied in compounds having a higher flame retardancy performance.

[0032] Also preferably, the amount of flame retardant is at most 90 parts by weight, more preferably at most 85 parts by weight and most preferably at most 80 parts by weight, relative to the total amount of polyamide of 100 parts by weight. A lower minimum amount of flame retardant is advantageously applied in compounds having an increased toughness and impact resistance and/or flowability. The optimal amount can in principle be determined experimentally by a person skilled in the art of formulating polyamide compounds through systematic research.

[0033] In the compound according to the invention, next to the flame retardant also additives contributing to the flame retardancy behaviour may be present.

[0034] Suitable additives for this purpose are flame retardant synergists, charring agents, dripping modifiers and anti-dripping components.

[0035] Suitable synergists are, for example Group IIA and IIB metal borates, [like zinc borate, for example Firebrake® ZB (Borax Inc., USA)] and metal oxides and hydroxides [such as magnesium hydroxide, aluminium hydroxide, iron oxide, zinc oxide, calcium oxide and analogous substances].

[0036] Suitable carbon-forming substances are, for example, polyphenylene ether and polycarbonate.

[0037] Suitable additives that can modify the dripping behaviour are, for example, fluoropolymers such as polytetrafluoroethylene.

[0038] Suitable anti-drip components are, for example, polymers prepared from monomers comprising at least an ethylenically unsaturated monomer and at least a carboxylic acid containing monomer. Preferably, the monomers comprise ethylene as the ethylenically unsaturated monomer.

[0039] The compound according to the invention may also contain other additives known to a person skilled in the

art that are customarily used in polymer compounds, provided that they do not essentially detract from the invention. Such other additives include, in particular, fillers, pigments, processing aids, for example mould release agents, agents accelerating crystallization, nucleating agents, lubricants, softeners, UV and heat stabilizers and the like. In particular, the compound according to the invention contains an inorganic filler or reinforcing agent. Suitable for use as an inorganic filler or reinforcing agent are all the fillers known to a person skilled in the art. Suitable reinforcing agents are, for example glass fibres, metal fibres, graphite fibres, aramide fibres, glass beads, aluminium silicates, asbestos, mica, clay, calcined clay and talcum.

[0040] Typically, the polyamide compound according to the invention consists of:

- a) 70-99.9 parts by weight of a polyamide polymer having a weight-average molecular weight of at least 10.000 g/mol
- b) 0.1-30 parts by weight of a polyamide oligomer having a molecular weight of at most 7500 g/mol, whereby the total amount of a) + b) is 100 parts by weight
- c) 1-100 parts by weight of a halogen-free phosphorous containing flame retardant
- d) 0-100 parts by weight of a reinforcing agent
- e) 0-25 parts by weight of at least one other component.

[0041] The invention also relates to a process for preparing a flame retardant polyamide compound according to the invention by melt-mixing of a composition comprising a polyamide polymer having a weight average molecular weight of at least 10.000 g/mol, a polyamide oligomer with a molecular weight below 7500 g/mol and a halogen-free phosphorous containing flame retardant. With the process according to the invention a polyamide compound with lower total burning times in flame retardancy tests can be prepared than with the corresponding process comprising melt-mixing of a polyamide composition not comprising a polyamide oligomer. The compound resulting from the inventive process also shows, as an additional advantage, at least retained mechanical properties, whereas moulded products prepared from said inventive compounds exhibit a good outer appearance, even with a polyamide oligomer content above 10 parts by weight, and/or with a content of flame retardant above 25 parts by weight, relative to 100 parts by weight of the polyamide polymer.

[0042] Preferred embodiments of the inventive process directly relate to the preferred embodiments of the polyamide compounds according to the invention described above.

[0043] The process according to the invention can be carried out in a melt-mixing apparatus, for which any melt-mixing apparatus known to the man skilled in the art of preparing polymer compounds by melt-mixing can be used. Suitable melt-mixing apparatus are, for example, kneaders, Banburry mixers, single screw extruders and double screw extruders. Melt-mixing is typically performed at a processing temperature, which is above the melting temperature of the high molecular weight polyamide, thereby forming a polymer melt.

[0044] In the process according to the invention for preparing a polyamide compound, comprising a polyamide polymer, a polyamide oligomer and a halogen-free phosphorous containing flame retardant as constituting components amongst possible other constituting components, the constituting components are fed to a melt-mixing apparatus and melt-mixed in that apparatus. The constituting components, i.e. polymer and oligomer, may be fed simultaneously as a powder mixture or granule mixer, also known-as dry-blend, or may be fed separately. Polyamide polymer and polyamide oligomer may also be fed separately.

[0045] In a preferred emodiment of the inventive process, the halogen-free phosphorous containing flame retardant is a melamine based phosphorous compound. The advantage is that the melt-mixing may be performed at higher temperature and a higher melting polyamide may be used with retention of a good flame retardancy performance.

[0046] In another preferred embodiment, the polyamide polymer is a high temperature engineering polyamide, defined by a melting temperature of at least 260°C. More preferable, the melting temperature is at least 270°C, even more preferable at least 280°C, and most preferably at least 290°C. The higher the melting temperature of the polyamide polymer, the more pronounced the effect of addition of the polyamide oligomer in the melt-mixing process on retention and/or improvement in mechanical properties, and/or flame retardancy and/or surface appearance of the compound formed, is.

[0047] Also preferably, the polyamide oligomer has a melting temperature of at most 20°C above, more preferable at most 10°C above to the melting temperature of the polyamide polymer, and even more preferable at most equal to the melting temperature of the polyamide polymer. The advantage of a polyamide oligomer with a melting temperature that does not exceed the melting temperature of the polyamide polymer too much or even better with a melting temperature that does not exceed the melting temperature of the polyamide polymer at all, is that the processing temperature necessary for the melt-mixing process can be kept as low as or can even be reduced compared to the process without polyamide oligomer and further reduces the effects of degradation of the flame retardant.

[0048] The process according to the invention is in particular advantageously applied for preparing a flame retardant polyamide compound comprising a reinforcing agent. Preparation of a flame retardant polyamide compound comprising

15

20

5

10

25

35

30

40

45

50

55

a reinforcing agent generally suffers even more from deterioration of the flame retardant, or the negative effects thereof upon the properties of the polyamide compound, than non reinforced compounds. With the process according to the invention flame retardant reinforced polyamide compounds can be prepared that show a substantially reduced negative effect of the flame retardant on the mechanical properties of the compound.

[0049] Preferably, glass fibres are chosen as the reinforcing agent. With the process according to the invention, also compounds with higher glass content, without a noticeable effect on the flame retardant and without significant decline in flame retardant properties, can be prepared.

[0050] The invention also relates to the use of a flame retardant polyamide compound according to the invention for preparing a molded part. The advantage is that processing of the compound can be done at lower processing temperatures and/or lower pressure than corresponding compounds without a polyamide oligomer. Furthermore, the parts have a better flame retardancy and improved mechanical properties.

[0051] The invention furthermore relates to a molded part obtainable by melt-processing of a polyamide compound

[0052] Suitable melt-processes are, for example, injection moulding, blow-moulding and pultrusion. The molded part may be, for example, a part for automotive use, or electrical or electronic use. Examples of such parts include, for 15 example, connectors and switches.

[0053] The invention will now be elucidated with reference to the following examples, but is not limited thereto.

Materials	
PP-A	Polyamide polymer: Stanyl® KS200 (ex DSM, The Netherlands): Polyamide-4,6 polymer, Mw = 36000, viscosity number (formic acid) = 160; T _{melt} = 295°C.
PO-A MPP-200	Polyamide oligomer: (ex DSM, The Netherlands): M _w = 2,000, T _{melt} = 288°C. Melamine polyphosphate: Melapur 200 (ex DSM, The Netherlands): nitrogen content 42-44 weight %.
PMP-100 Glass fibres	Melamine polyphosphate; (Nissan Chemical Industries Ltd); phosphor content 14.5 weight %. Standard glass fibres for polyamide compounds; average fiber diameter 10 µm.

Determination of the physical properties

[0054]

30

35

40

45

55

10

Viscosity number: determined in formic acid, according to ISO 307

Tensile strength: determined at 23°C and 5 mm/min, according to ISO 527

Elongation at break: determined at 23°C and 5 mm/min, according to ISO 527

Notched Izod: determined at 23°C according to ISO 180/1A

Molecular weight: determined with the aid of standard GPC techniques

Melting point: determined with the aid of DSC (2nd run, 10°C/min.).

Flame retardancy: determined according Underwriters Laboratories test method UL 94, using 0.8 mm test bars, conditioned for 48 hours at 23°C, 50% relative humidity, respectively for 168 hours at 70°C.

Bulk density: determined according ASTM D 1895-96 Test Method A.

Preparation of polyamide compounds

Example I and Comparative Experiment A

[0055] Polyamide compounds according Example I, comprising a combination of PP-A and PO-A, and Comparative Experiment A, only comprising PP-A as the polyamide, both comprising 30 weight % of MPP-200, relative to the total weight of the compound (see Table I) were prepared by melt-mixing the constituent components on a Werner & Pfleiderer ZSK-40 twin screw extruder using a 300°C flat temperature profile. The constituents were fed via a hopper, glass fibers were added via a side feed. Throughput was 60 kg/h and screw speed was 250 rpm. The polymer melt was degassed at the end of the extruder. The melt was extruded into strands, cooled and chopped into granules. [0056] The granules were injection moulded into test bars according to ISO 527/1A multipurpose specimens and UL

94 test bars of 0.8 mm thick. The test bars were used to measure the flame retardant properties and mechanical properties of the compounds, the results of which have been reported in Table I.

Table II.

Composition and test results of E	xample I and Comparative Exp	periment A	
	Comparative Experiment A	Example I	
Components(weight %):			
PP-A	40	34	
PO-A	·	6	
MPP-200	30	30	
Glass fibre	. 30	30	
Throughput in kg/h	60	60	
Strand formation	Very bad	o.k.	
Notched Izod (kJ/m²)	8.6	9.0	
Tensile strength (MPa)	101	110	
Elongation at break (%)	0.8	1.0	
UL 94 V flammability properties:			
0.8mm, 70°C/168h	. ,		
V0- Classification:	100%	100%	
Total burning time	23	21	
0.8mm, 23°C/50%RH/48h			
V0-Classification	100%	100%	
Total burning time	. 19	10	

Example II and Comparative Experiment B.

[0057] Example II and Comparative Experiment B represent compounds similar to Example I and Comparative Experiment A, except that PMP-100 was used as the flame retardant. The detailed processing conditions, applied for the preparation of the compounds are the same as described for example I and comparative experiment A. Test results obtained from injection-molded parts prepared from respective granulates have been collected in Table II.

Table II.

	Table II.	
Composition and test results of E	xample II and Comparative Ex	periment B
	Comparative Experiment B	Example II
Components (weight %):		
PP-A	40	34
PO-A		6
PMP-100	30	30
Glass fibre	30 ·	30
Throughput (kg/h)	60	60
Strand formation	o.k.	o.k.
Tensile strength (MPa)	109	134
Elongation at break (%)	1.0	1.3
UL 94 V flammability properties:		
0.8mm, 70°C/168h	·	·
V0- classification:	80%	100%
Total burning time (sec)	47	34
0.8mm, 23°C/50%RH/48h		
V0-classification	0%	60%
Total burning time (sec)	90	45

Claims

5

10

- 1. Flame retardant polyamide compound comprising a polyamide polymer having a weight-average molecular weight of at least 10.000 g/mol, a halogen-free flame retardant and a polyamide oligomer having a weight-average molecular weight of at most 7500, characterized in that the halogen-free flame retardant is a halogen-free phosphorous containing flame retardant.
 - 2. Compound according to claim 1, wherein the polyamide oligomer is a polyamide with a melting temperature of at least 260°C.
 - Compound according to claim 1 or 2, wherein the polyamide oligomer is present in an amount of 0.1-30 weight %, relative to the total weight of polyamide.
- Compound according to any of claims 1-3, wherein the halogen-free phosphorous containing flame retardant is a
 melamine based phosphorous compound.
 - 5. Compound according to any of claims 1-4, wherein phosphorous containing flame retardant is present in an amount between 1 and 100 parts by weight, relative to a total amount of polyamide of 100 parts by weight.
- 6. Process for preparing a compound according to any of claims 1-5 comprising melt-mixing of a polyamide composition comprising a polyamide polymer having a weight-average molecular weight of at least 10.000 g/mol, a polyamide oligomer having a weight-average molecular weight of at most 7500, and a halogen-free phosphorous containing flame retardant.
- Process according to claim 6, wherein the polyamide polymer is a polyamide with a melting temperature of at least 260°C.
 - Process according to any of claims 6-7, wherein the polyamide oligomer has a melting temperature of at most 20°C above the melting temperature of the polyamide polymer.
 - 9. Process according to any of claims 1-5, wherein the polyamide compound comprises a reinforcing component.
 - 10. Use of a polyamide compound according to any of claims 1-6 for the preparation of a molded part.
- 35 **11.** Molded part obtainable by melt-processing of a polyamide compound according to any of claims 1-6.

40

30

45

50



PARTIAL EUROPEAN SEARCH REPORT

Application Number

which under Rule 45 of the European Patent Convention EP 02 08 0410 shall be considered, for the purposes of subsequent proceedings, as the European search report

	DOCUMENTS CONSIDE	RED TO BE RELEVANT]
Category	Citation of document with inco	dication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X Y	EP 0 416 430 A (BASI 13 March 1991 (1991- * claims 1,2,8-10 *	1,3,5-7, 9-11	C08K5/49 C08K5/3492 C08L77/00	
Y	WO 00 75233 A (CREVE ;KONING CORNELIS EME KO) 14 December 2006 * claims 1-3,6 *	E (NL): DSM NV (NL):	1-3,5-11	·
	US 5 541 267 A (AKKA 30 July 1996 (1996-6 * claims 1,7,11 *	APEDDI MURALI K ET AL) 07-30)	1-11	
\	EP 0 498 977 A (MITS POLYCHEMICAL) 19 Aug * claims 1-3,8,15 *	UI DU PONT Just 1992 (1992-08-19)	1-11	
	·			
				TECHNICAL FIELDS SEARCHED (Int.CL.7)
				C08K C08L
he Search	PLETE SEARCH n Division considers that the present ap with the EPC to such an extent that a nout, or can only be carried out partially,	plication, or one or more of its claims, does/ neaningful search into the state of the art ca for these dalms.	(do Innot	* v
	rched completely :			-
	rched incompletely :			
	the limitation of the search:			
,		•		
				1
	Ptace of search THE HAGUE	Oate of completion of the search		Examiner
		27 May 2003	Rose	
X : particu Y : particu docum A : techno O : non-w	EGORY OF CITED DOCUMENTS alarly relevant if taken alone alarly relevant if combined with another ent of the same category ological background inten disclosure ediate document	T: theory or principle E: earlier patent doc after the filing date D: document cited in L: document cited fo	ument, but publish the application rother reasons	ed on, or

EPO FORM 1503 03.82 (P04C07)



INCOMPLETE SEARCH SHEET C

Application Number

EP 02 08 0410

Claim(s) searched incompletely: 1-11

Reason for the limitation of the search:

Present claims 1-11 relate to a polyamide compound defined by reference to the following parameters:

P1 : (polyamide polymer having) a weight average molecular weight of at least 10 000 g/mol

P2 : (polyamide oligomer having) a weight average molecular weight of at most 7500 g/mol

The use of these parameters in the present context is considered to lead to a lack of clarity within the meaning of Article 84 EPC. It is impossible to compare the parameters the applicant has chosen to employ with what is set out in the prior art. The lack of clarity is such as to render a meaningful complete search impossible. Consequently, the search has been restricted to the parts relating to the polyamide compound mentioned in claim 1, but without the parameters conditions P1 and P2, that is to say claim 1 such as following: "Flame retardant polyamide compound comprising a polyamide polymer, a halogen-free flame retardant and a polyamide oligomer, characterized in that the halogen-free flame retardant is a halogen-free phosphorous containing flame retardant".

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 02 08 0410

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

27-05-2003

DE 59001198 D1 19-05-199 EP 0416430 A1 13-03-199 ES 2042162 T3 01-12-199 WO 0075233 A 14-12-2000 NL 1012234 C2 06-12-200 AT 232551 T 15-02-200 AU 5114700 A 28-12-200 CN 1368994 T 11-09-200 DE 60001408 D1 20-03-200 EP 1183310 A1 06-03-200 JP 2003501538 T 14-01-200 JP 2003501538 T 14-01-200 US 2002103287 A1 01-08-200 US 5541267 A 30-07-1996 AU 693491 B2 02-07-199 AU 1100795 A 06-06-199 CA 2176009 A1 26-05-199 DE 69413247 T2 25-03-199 DE 69413247 T2 25-03-199 DE 69413247 T2 25-03-199 DF 9502478 T 11-03-199 JP 3074448 A 29-03-199 CA 2036059 A1 12-08-199	Patent docume cited in search re		Publication date		Patent family member(s)	Publication date
MT 232551 T 15-02-206 AU 5114700 A 28-12-206 CN 1368994 T 11-09-206 DE 60001408 D1 20-03-206 EP 1183310 A1 06-03-206 JP 2003501538 T 14-01-206 WO 0075233 A1 14-12-206 US 2002103287 A1 01-08-206 US 2002103287 A1 01-08-206 AU 1100795 A 06-06-199 CA 2176009 A1 26-05-199 DE 69413247 D1 15-10-199 DE 69413247 T2 25-03-199 DE 69413247 T2 25-03-199 DP 2744138 B2 28-04-199 JP 2744138 B2 28-04-199 JP 2744138 B2 28-04-199 JP 9502478 T 11-03-199 WO 9514059 A1 26-05-199 EP 0498977 A 19-08-1992 JP 2523887 B2 14-08-199 CA 2036059 A1 12-08-1999 CA 2036059 A1 12-08-1999	EP 0416430	Α	13-03-1991	DE EP	59001198 D1 0416430 A1	07-03-1991 19-05-1993 13-03-1991 01-12-1993
AU 11007795 A 06-06-199 CA 2176009 A1 26-05-199 DE 69413247 D1 15-10-199 DE 69413247 T2 25-03-199 EP 0729490 A1 04-09-199 JP 2744138 B2 28-04-199 JP 9502478 T 11-03-199 W0 9514059 A1 26-05-199 EP 0498977 A 19-08-1992 JP 2523887 B2 14-08-199 JP 3074448 A 29-03-199 CA 2036059 A1 12-08-199	WO 0075233	A	14-12-2000	AT AU CN DE EP JP WO	232551 T 5114700 A 1368994 T 60001408 D1 1183310 A1 2003501538 T 0075233 A1	06-12-2000 15-02-2003 28-12-2000 11-09-2002 20-03-2003 06-03-2002 14-01-2003 14-12-2000 01-08-2002
JP 3074448 A 29-03-199 CA 2036059 A1 12-08-199	US 5541267	A	30-07-1996	AU CA DE DE EP JP	1100795 A 2176009 A1 69413247 D1 69413247 T2 0729490 A1 2744138 B2 9502478 T	02-07-1998 06-06-1995 26-05-1995 15-10-1998 25-03-1999 04-09-1996 28-04-1998 11-03-1997 26-05-1995
10 00 100	EP 0498977	A	19-08-1992	JP CA EP	3074448 A 2036059 A1 0498977 A1	14-08-1996 29-03-1991 12-08-1992 19-08-1992 11-05-1993

FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

THIS PAGE BLANK (USPTO)